

## **JOHNSTON SELF-EVALUATION GUIDE**

Self-evaluation is an essential part of Johnston contracting for two reasons: 1) it allows students to make their own assessments about the work they have done, thus giving them a real sense of ownership of their studies; 2) it provides faculty with crucial feedback for evaluating students as well as for improving their effectiveness as facilitators/instructors. Self-evaluation institutionalizes the balanced relationship between students and faculty, which is central to a Johnston education.

The evaluation form is divided into two sections. In the first section, students should comment on their own learning. In the second section, students should comment on the effectiveness of the faculty member. In both sections, students are encouraged to talk about group processes as they relate to individual learning. **Make specific reference to the course contract and how it was fulfilled.** Below are some questions to help frame a self-evaluation; feel free to answer others which are more specific to each particular course contract.

### **PART I / Student Self-Evaluation**

What objectives did you set for yourself and the group? How successfully did you accomplish these goals?

List your accomplishments in the course.

What learning devices worked best for you?

Did you participate in the formation of the group process and the course design?

How would you assess your participation in class discussions?

What books, people, or other learning devices were especially valuable?

What was your most/least successful or meaningful project and why?

### **PART II / Faculty Evaluation**

Did the faculty provide sufficient breadth and depth to insure quality of learning? Did the faculty have an appropriate grasp of the subject?

How sensitive was the faculty to individual needs and the needs of the class?

How was the faculty responsive to criticism?

How do you assess the faculty's ability to organize and facilitate the course?

Was the faculty open to contract negotiation?